Home Search Collections Journals About Contact us My IOPscience

Temperature-dependent nuclear magnetic resonance study of 133 Cs in an LiCsSO₄ single crystal

This article has been downloaded from IOPscience. Please scroll down to see the full text article.

1999 J. Phys.: Condens. Matter 11 8141

(http://iopscience.iop.org/0953-8984/11/41/316)

View the table of contents for this issue, or go to the journal homepage for more

Download details: IP Address: 171.66.16.214 The article was downloaded on 15/05/2010 at 13:28

Please note that terms and conditions apply.

Temperature-dependent nuclear magnetic resonance study of ¹³³Cs in an LiCsSO₄ single crystal

Ae Ran Lim[†]_{||}, Sung Ho Choh[‡] and Se-Young Jeong§

[†] Department of Physics, Jeoniu University, Jeoniu 560-759, Korea

‡ Department of Physics, Korea University, Seoul 136-701, Korea

§ Department of Physics, Pusan National University, Pusan 609-735, Korea

E-mail: aeranlim@hanmail.net

Received 17 May 1999, in final form 10 August 1999

Abstract. Temperature dependence of the ¹³³Cs nuclear magnetic resonance in an LiCsSO₄ single crystal grown by the slow evaporation method has been investigated by employing an FT NMR spectrometer. From the experimental data, the quadrupole coupling constant, asymmetry parameter and direction of the EFG tensor were determined in the temperature range from 180 K to 400 K. The ¹³³Cs NMR spectra shows a phase transition at 200 K which is connected with the formation of three kinds of ferroelastic domain. The obtained results are explained by the existence of these three domains, rotated with respect to each other by 60° around the *c*-axis. The nuclear quadrupole coupling constant and asymmetry parameter obtained as functions of temperature for ¹³³Cs in an LiCsSO₄ single crystal are described by the linear equations $e^2q Q/h = 178 + 1.05 \times 10^{-1} (T - T_r)$ kHz and $\eta = 0.9221 - 2.33 \times 10^{-4} (T - T_r)$, respectively, in the temperature range 180–400 K.

1. Introduction

LiCsSO₄ single crystals undergo a phase transition from the paraelastic phase with orthorhombic structure at room temperature to the ferroelastic phase with monoclinic structure below $T_C = 202$ K [1, 2]. The space group of the high temperature phase and that of the low temperature phase are *Pcmn* and *P*112₁/*n*, respectively, without cell multiplication. The main changes in the structure as a result of this transition are described by rotation of the SO₄ tetrahedra about the *c*-axis [3]. LiCsSO₄ crystals have been studied by x-ray diffraction [4, 5], Raman spectroscopy [6–8], EPR [9–13], NMR [14–16] and optical properties [17].

In this paper, we discuss the ¹³³Cs nuclear magnetic resonance (NMR) in an LiCsSO₄ single crystal grown by the slow evaporation method. The temperature dependence of the quadrupole coupling constant, e^2qQ/h , the asymmetry parameter, η , and the direction of the principal axes of the electric field gradient (EFG) tensor of ¹³³Cs (I = 7/2) were determined with a pulse NMR spectrometer. Based on these results, we examine the phase transition and the ferroelastic property of an LiCsSO₄ single crystal.

2. Crystal structure

LiCsSO₄ is an orthorhombic structure with four molecular units per unit cell. The space group is *Pcmn* and the point group is *mmm*, and the lattice constants a = 9.456 Å, b = 5.456 Å and

^{||} To whom all correspondence should be addressed.

^{0953-8984/99/418141+07\$30.00 © 1999} IOP Publishing Ltd



Figure 1. (a) Projections of the atomic positions in the orthorhombic Pcmn phase of LiCsSO₄ onto a (001) plane. (b) Projections of the atomic positions in the monoclinic $P2_1/n$ phase of LiCsSO₄ onto a (001) plane.

c = 8.820 Å. The four Li and Cs atoms in a unit cell are crystallographically equivalent. The oxygen coordination around each Cs is 11-fold, and the Li⁺ ion has a tetrahedral coordination with Li–O distance ranging from 1.849 to 1.954 Å [4]. Figure 1(a) shows the projections of the atomic positions in the *Pcmn* phase onto a (001) plane.

The orthorhombic to monoclinic transition is characterized by the disappearance of the *b*-mirror symmetry and an ordering of the sulphate groups. At 163 K, the space group is monoclinic $P2_1/n$ and the point group is 2/m [12]. The lattice parameters are a = 9.379 Å, b = 5.423 Å, c = 8.834 Å and $\gamma = 89.75^{\circ}$. In the monoclinic phase, there are still four molecular units in a unit cell, and all the atoms are crystallographically equivalent. Figure 1(b) shows the projections of the atomic positions at 163 K onto a (001) plane.

3. Experimental procedure

Clear single crystals of LiCsSO₄ were grown at room temperature by slow evaporation from aqueous solution containing equimolar amounts of Li₂SO₄ · H₂O and Cs₂SO₄. The transparent and colourless single crystals prepared were about $5 \times 5 \times 6$ mm³. The crystals are hexagonal in shape and display a 60° twin structure. As samples for measurements, untwinned parts were selected from these crystals. The orientation of the crystal was identified using an optical polarizing microscope and the x-ray Laue method. The angular dependence of the NMR spectra was measured on the crystallographic *ab*-, *bc*- and *ca*-planes, respectively.

Nuclear magnetic resonance signals of ¹³³Cs in the LiCsSO₄ single crystal were measured using a Varian 400 FT NMR spectrometer at the Korea Basic Science Institute in Taejon. The static magnetic field was 9.4 T and the central rf frequency was set at $\omega_0/2\pi = 52.484$ MHz. The NMR spectra of ¹³³Cs were recorded with a sequence of one 90° pulse, 60 scans and a repetition time of 10 s. A ring-down delay time of 3 μ s was used to remove the effect of the pulse. The temperature-dependent NMR measurements were obtained using a previously reported method [18].



Figure 2. ¹³³Cs NMR spectra in the LiCsSO₄ crystal: (a) room temperature; (b) 180 K.

4. Experimental results and analysis

The seven-line structure is a result of the quadrupole interaction of the ¹³³Cs (I = 7/2) nucleus. A typical NMR spectrum of ¹³³Cs is shown in figure 2(a). Rotation patterns in three mutually perpendicular directions are required to determine the quadrupole interaction completely. The rotation patterns of ¹³³Cs NMR spectra measured in the three crystallographic planes at room temperature are displayed in figure 3. Because the resonance frequency of the central line is almost constant and the spacings between adjacent lines are equal, the first-order perturbation of the nuclear electric quadrupole term with respect to the Zeeman term may be sufficient for analysis [19, 20]. The satellite resonance lines show an angular dependence of $3 \cos^2 \theta - 1$, where the polar angle θ is the direction of the magnetic field with respect to the *c*-axis, whereas the central lines are angle independent. At this time, the satellite resonance lines did not



Figure 3. The rotation pattern of 133 Cs NMR measured in the *ab*-, *bc*- and *ca*-plane at room temperature.

show maximum or minimum separation in the crystallographic axis. The parameters were determined by a least-squares fit to the experimental data. The nuclear quadrupole coupling constant and asymmetry parameter of ¹³³Cs in a LiCsSO₄ crystal were determined as 181 kHz and 0.92 at room temperature. The direction of the principal EFG tensor for the Cs ion at 300 K is represented with the Eulerian angles $\Phi = 30^{\circ}$, $\Theta = 10^{\circ}$ and $\Psi = 30^{\circ}$. These results are inconsistent with the previous report [14]. The geometric angles of the Cs–O bonds in the Cs–O polyhedra is approximately 30° to the *a*-axis as shown in figure 1. These directions are reasonably close to Eulerian angles $\Phi = 30^{\circ}$.

In order to examine the temperature dependence of $e^2 q Q/h$ and η for ¹³³Cs in the LiCsSO₄ single crystal, the resonance spectra were measured at several temperatures in the range 180-400 K including the phase transition temperature. Originally, the seven-line structure was a result of the quadrupole interaction of the ¹³³Cs (I = 7/2) nucleus. The ¹³³Cs NMR spectrum consists of only seven lines for all orientations of the crystal in the external magnetic field in the 205-400 K temperature range. However, the ¹³³Cs NMR line splits into three set lines below the transition point of 200 K as shown in figure 2(b). Here, the signal for one set showed very strong intensity, while another signal showed weak intensity. One signal labelled by closed diamonds in these three groups is shown to be of very weak intensity. One of the rotation patterns of the ¹³³Cs NMR spectra measured in the *ab*-plane at 180 K is displayed in figure 4. The ¹³³Cs NMR lines were a superposition of the resonance lines of one domain and the lines rotated by a 60° angle with respect to the *c*-axis. The domain wall makes an angle of 60° and 120° with respect to the *a*-axis. The obtained results can be explained by the existence of three kinds of ferroelastic domain. This indicates that the LiCsSO4 crystal below T_C has characteristics of a twin structure crystal. The emergence of three sets of ¹³³Cs resonance lines indicates that a ferroelastic phase transition takes place to a new phase with lower symmetry than the orthorhombic. From the experimental data, three parameters were observed with the same magnitude of the quadrupole coupling constant and the asymmetry parameter. The nuclear quadrupole coupling constant and asymmetry parameter of ¹³³Cs in



Figure 4. The rotation patterns of 133 Cs NMR measured in the *ab*-plane at 180 K.



Figure 5. Temperature dependence of the nuclear quadrupole coupling constant, $e^2 q Q/h$, for ¹³³Cs in an LiCsSO₄ single crystal.

an LiCsSO₄ crystal were determined to be 165 kHz and 0.953 at 180 K. The principal axes X, Y and Z of the EFG tensor for the Cs ion are found to lie along the Eulerian angles above T_C .

The ¹³³Cs NMR spectra had only continuous quantitative changes in the quadrupole splitting distances. The temperature dependence of $e^2 q Q/h$ for the ¹³³Cs nucleus is shown in figure 5 as open circles. The value of $e^2 q Q/h$ was found to increase as a function of increasing temperature, where the experimental data are described by the linear equation $e^2 q Q/h = 178 + 1.05 \times 10^{-1} (T - T_r)$ kHz, as represented by the solid line. Here, *T* (K) is temperature and T_r (K) is room temperature. Generally, $e^2 q Q/h$ decreases as a function of



Figure 6. Temperature dependence of the asymmetry parameter, η , for ¹³³Cs in an LiCsSO₄ single crystal.

increasing temperature in many substances: that is, the slope $d\nu_Q/dT$ is negative. Conversely, however, $e^2 q Q/h$ for ¹³³Cs in LiCsSO₄ increases as the temperature increases. Figure 6 shows the temperature dependence of η , which decreases as the temperature increases. The solid line is the linear fit of η with a linear equation $\eta = 0.9221 - 2.33 \times 10^{-4}(T - T_r)$. This trend is similar to the temperature dependence of the ⁹Be NMR in BeAl₂O₄ single crystals previously reported [21].

5. Discussion and conclusion

The phase transition by ¹³³Cs NMR in an LiCsSO₄ single crystal grown by the slow evaporation method was investigated by employing a Varian 400 FT NMR spectrometer. From the experimental data, the quadrupole coupling constant, asymmetry parameter and direction of the EFG tensor were determined in the temperature range from 180 K to 400 K. In the temperature range between 180 K and 400 K, which includes the phase transition, there is no abrupt change in the parameters of the ¹³³Cs nucleus. The principal axes of the EFG tensor for the Cs ion below T_C are same as above T_C . According to the ⁷Li NMR results in LiCsSO₄ single crystals, in the temperature range between 140 K and 380 K, there is no abrupt change in the parameter of ⁷Li NMR. Only the e^2qQ/h and η of the ⁷Li nucleus decrease as temperature increases. However, the principal axes of the ⁷Li EFG tensor below T_C are not the same as above T_C [16]. A change in the EFG tensor in ⁷Li at T_C indicates that the LiO₄ tetrahedra are significantly affected during this transition. This is consistent with the observation that larger changes in the EFG tensor are obtained at the site of ⁷Li than at ¹³³Cs at T_C . We conclude that the Cs ion plays only a minor role in this transition mechanism.

The ¹³³Cs NMR spectra demonstrate the occurrence of a phase transition at 200 K which is connected with the formation of these three domains. The obtained results can be explained by the existence of three kinds of ferroelastic domain, rotated with respect to each other by 60° around the *c*-axis. The quadrupole coupling constant of ¹³³Cs in an LiCsSO₄ crystal increases as the temperature increases. However, the asymmetry parameter decreases with increasing temperature. The nuclear quadrupole coupling constant and asymmetry parameter obtained as functions of temperature for ¹³³Cs in LiCsSO₄ single crystals can be described by the linear equations $e^2 q Q/h = 178 + 1.05 \times 10^{-1} (T - T_r)$ kHz and $\eta = 0.9221 - 2.33 \times 10^{-4} (T - T_r)$, respectively, in the temperature range 180–400 K.

Acknowledgments

This work was supported by the Basic Science Research Institute Programme, Ministry of Education, 1999, project No BSRI-99-2410 (2411) and by the Korea Science and Engineering Foundation (KOSEF) through the Research Centre for Dielectric and Advanced Matter Physics (RCDAMP) at Pusan National University (1997–2000).

References

- [1] Pietraszko A, Tomaszewski P E and Lukaszewicz K 1981 Phase Transitions 2 141
- [2] Chary B R, Bhat H L, Chandrasekhar P and Narayanan P S 1985 Pramana 24 545
- [3] Zamkov A V and Anistratov A T 1982 Sov. Phys.-Solid State 24 873
- [4] Kruglik A I, Simonov M A, Zhelezin E P and Belov N V 1979 Sov. Phys.-Dokl. 24 596
- [5] Niwata A and Itoh K 1995 J. Phys. Soc. Japan 64 4733
- [6] Rao T R, Bansal M L, Sahni V C and Roy A P 1976 Phys. Status Solidi b 75 k31
- [7] Morell G, Devanarayanan S and Katiyar R S 1991 J. Raman Spectrosc. 22 529
- [8] Shashikala M N, Chandrabhas N, Jayaram K, Jayaraman A and Sood A K 1994 J. Phys. Chem. Solids 55 107
- [9] Morais P C, Ribeiro G M and Chaves A S 1984 Solid State Commun. 52 291
- [10] Huang S J and Yu J T 1987 Solid State Commun. 63 745
- [11] Tsu Y J 1988 J. Phys. C: Solid State Phys. 21 759
- [12] Yu J T, Chou S Y and Huang S J 1988 J. Phys. Chem. Solids 49 289
- [13] Misra S K and Misiak L E 1993 Phys. Rev. B 48 13 579
- [14] Holuj F 1985 Ferroelectrics 65 55
- [15] Holuj F 1986 Ferroelectrics 67 103
- [16] Lim A R and Jeong S Y 1998 J. Phys.: Condens. Matter 10 9841
- [17] Czajkowski M, Drozdowski M and Kozielski M 1988 Phys. Status Solidi a 110 437
- [18] Lim A R, Choh S H and Jeong S Y 1996 J. Phys.: Condens. Matter 8 4589
- [19] Abragam A 1961 The Principles of Nuclear Magnetism (Oxford: Oxford University Press) ch VII
- [20] Slichter C P 1989 Principles of Magnetic Resonance (Berlin: Springer) ch 10
- [21] Yeom T H, Lim A R, Choh S H, Hong K S and Yu Y M 1995 J. Phys.: Condens. Matter 7 6117